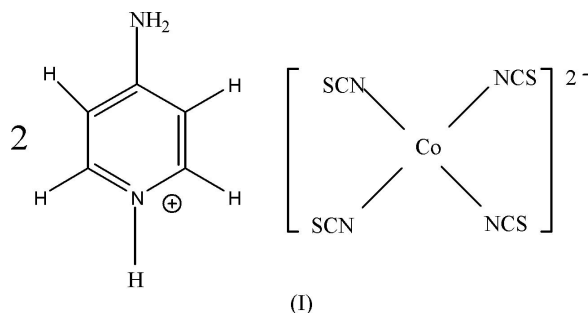


Jing Min Shi,* Feng Xia Zhang,
Jian Jun Lu and Lian Dong LiuDepartment of Chemistry, Shandong Normal
University, Jinan 250014, People's Republic
of ChinaCorrespondence e-mail:
shijingmin@beelink.com

Key indicators

Single-crystal X-ray study
 $T = 298$ K
Mean $\sigma(\text{C}-\text{C}) = 0.009$ Å
 R factor = 0.063
 wR factor = 0.149
Data-to-parameter ratio = 16.1For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Bis(4-aminopyridinium) tetrathiocyanato-
cobaltate(II)The title structure, $(\text{C}_5\text{H}_7\text{N}_2)_2[\text{Co}(\text{NCS})_4]$, comprises discrete monovalent 4-aminopyridinium cations and divalent tetrathiocyanatocobaltate(II) anions. The cations and anions are linked *via* $\text{N}-\text{H}\cdots\text{S}$ hydrogen bonds [$\text{N}\cdots\text{S} = 3.264$ (4)–3.640 (6) Å] to form a three-dimensional framework.Received 3 May 2005
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Comment

The Co atom of the anion in the title salt, (I), is coordinated by four N atoms from four thiocyanate groups in a slightly distorted tetrahedral geometry (see Table 1). In the crystal structure, there are hydrogen bonds between NH groups and S atoms, in which the NH donor groups are from both amino groups and pyridine ring N atoms (see Table 2) and the resulting $\text{N}-\text{H}\cdots\text{S}$ hydrogen bonds form a three-dimensional framework (see Fig. 2). In addition, there are significant $\pi-\pi$ stacking interactions between neighbouring pyridine rings; the relevant distances are $\text{Cg1}\cdots\text{Cg1}^{\text{i}} = 3.548$ (3) Å and $\text{Cg1}\cdots\text{1}^{\text{i}}_{\text{perp}} = 3.306$ Å, and $\text{Cg2}\cdots\text{Cg2}^{\text{ii}} = 3.925$ (4) Å and $\text{Cg2}\cdots\text{2}^{\text{ii}}_{\text{perp}} = 3.368$ Å [symmetry codes: (i) $1-x, 1-y, -z$; (ii) $-x, 1-y, -z$; Cg1 and Cg2 are the centroids of the N6/C5–C9 and N7/C10–C14 rings, respectively; $\text{CgI}\cdots\text{J}_{\text{perp}}$ is the perpendicular distance from CgI to ring J].

Experimental

4-Aminopyridine (0.0574 g, 0.610 mmol) was added to an aqueous solution (15 ml) containing $\text{Co}(\text{ClO}_4)_2\cdot 6\text{H}_2\text{O}$ (0.1102 g, 0.301 mmol) and sodium thiocyanate (0.0511 g, 0.630 mmol), and the solution was stirred for a few minutes. Blue single crystals were obtained after the solution was allowed to stand at room temperature for four weeks.

Crystal data

 $(\text{C}_5\text{H}_7\text{N}_2)_2[\text{Co}(\text{NCS})_4]$
 $M_r = 481.50$
Monoclinic, $P2_1/c$
 $a = 14.118$ (2) Å
 $b = 9.1179$ (14) Å
 $c = 16.756$ (3) Å
 $\beta = 98.839$ (2)°
 $V = 2131.4$ (6) Å³
 $Z = 4$ $D_x = 1.501$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 2201
reflections
 $\theta = 2.6$ – 22.4 °
 $\mu = 1.21$ mm⁻¹
 $T = 298$ (2) K
Prism, blue
 $0.25 \times 0.09 \times 0.08$ mm

Data collection

Bruker SMART CCD diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.751$, $T_{\max} = 0.909$
 11075 measured reflections

3936 independent reflections
 2795 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.043$
 $\theta_{\text{max}} = 25.5^\circ$
 $h = -17 \rightarrow 17$
 $k = -11 \rightarrow 10$
 $l = -20 \rightarrow 16$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.063$
 $wR(F^2) = 0.149$
 $S = 1.05$
 3936 reflections
 244 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0635P)^2 + 1.614P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.71 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.37 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (\AA , $^\circ$).

Co1—N1	1.939 (4)	Co1—N3	1.951 (4)
Co1—N4	1.944 (4)	Co1—N2	1.956 (4)
N1—Co1—N4	109.84 (16)	N1—Co1—N2	107.78 (17)
N1—Co1—N3	115.66 (18)	N4—Co1—N2	114.37 (16)
N4—Co1—N3	101.76 (17)	N3—Co1—N2	107.56 (16)

Table 2

Hydrogen-bonding geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N6—H7 \cdots S1 ⁱ	0.86	2.54	3.264 (4)	142
N5—H4B \cdots S4 ⁱⁱ	0.86	2.75	3.540 (5)	154
N7—H13A \cdots S2 ⁱⁱⁱ	0.86	2.53	3.359 (7)	163
N8—H9A \cdots S3 ^{iv}	0.86	2.71	3.567 (6)	176
N8—H10B \cdots S2 ^v	0.86	2.79	3.640 (6)	170

Symmetry codes: (i) $x, 1+y, z$; (ii) $1-x, 1-y, -z$; (iii) $x-1, y, z$; (iv) $1-x, \frac{1}{2}+y, \frac{1}{2}-z$; (v) $x-1, \frac{1}{2}-y, z-\frac{1}{2}$.

All H atoms were placed in calculated positions and included in the final cycles of refinement using a riding-model approximation [$C-H = 0.93 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2_{\text{eq}}(\text{C})$; $N-H = 0.86 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2_{\text{eq}}(\text{N})$].

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Sheldrick, 2001) and PLATON (Spek, 2003); software used to prepare material for publication: SHELXTL.

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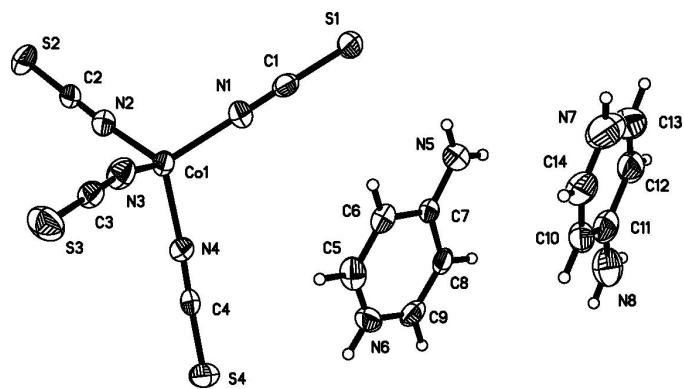


Figure 1

View of the title compound, showing the atom-numbering scheme. Ellipsoids are drawn at the 30% probability level. H atoms are shown as small spheres of arbitrary radii.

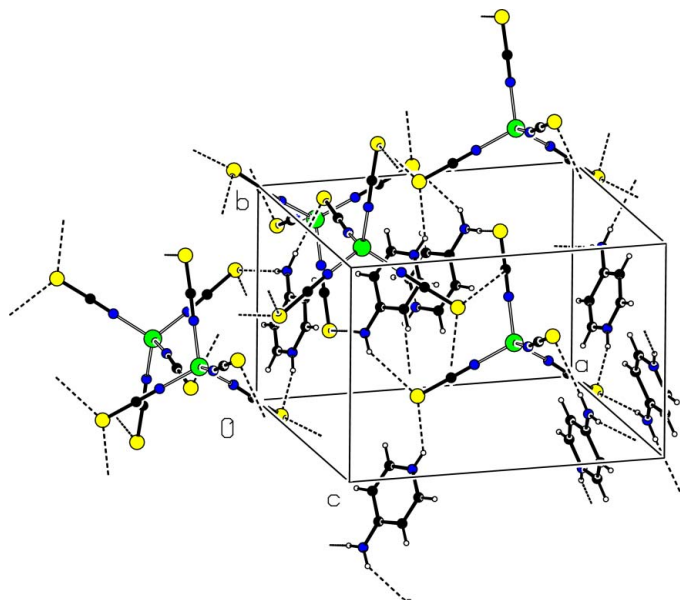


Figure 2

Packing diagram (Spek, 2003), showing $N-H \cdots S$ hydrogen bonds as dashed lines. Colour codes: green Co, yellow S, blue N and black C.

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